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Note: Not all of the  
reactions will occur. For  
those that do not, write  
no reaction. 1. A piece of  
copper is dropped into a  
container of water. No  
reaction 2. Liquid

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bromine is added to a container of sodium iodide crystals.  $\text{Br}_2(l) + 2\text{NaI}(s) \rightarrow 2\text{NaBr}(s) + \text{I}_2(s)$

3. An aluminum strip is immersed in a solution of silver nitrate.

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answers - Max Study

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SnCO tin(II) carbonate

sodium hydrogen

carbonate NaHCO 1. 2.

3. 4. 5. 6. 8. 9. 10. 11.

vanadium(V) oxide H<sub>2</sub>O

dihydrogen monoxide



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(NH) CO ammonium  
oxalate 4 224  
polonium(VI)  
thiocyanate Po(SCN)6  
tetraphosphorus  
decaoxide P O 4 10 12.  
13. 15. 16. 17. 18. 19.  
manganese(VII) oxide  
copper(II) dihydrogen  
phosphate

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predict and balance the  
following single  
replacement reactions.

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Since many of the  
transuranium elements  
were prepared at the  
Lawrence Radiation  
Center, University of  
California at Berkeley,  
California, U.S.A., it is  
logical to find the

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following names:

Americium—named after the Americas (after all, one must be patriotic!).

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Jane D Smith question  
requiring the  
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products of chemical  
reactions (the Equations  
section) five of these  
eight equations to answer

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Not all of the reactions  
will occur. For those that  
do not, write no reaction.  
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nanodiagnosics,  
nanotherapy, and  
nanorobotics are  
expected to revolutionize  
the future of medicine,  
leading to  
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highly effective targeted  
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